

Harvesting Light in a Changing Environment

Without exception, everything that happens in the living world needs energy to make it run – energy that comes ultimately from the sun. Our own lives depend on the accumulated legacy of a hundred million years of conversion of sunlight, compressed into the coal, oil and gas on which our style of civilization depends. The energy in our food was converted and stored recently, when the plants that we eat (even if at second hand) used sunlight for their growth. When you ride a bicycle, make no mistake that the wheels are driven by nuclear fusion, albeit fusion that happened some years ago and in a reactor at the safe distance of ninety three million miles. The wheels of your car go round for the same reason, but in that case the energy from the fusion reactor was accumulated and stored long ago, when ferns and amphibians by and large had things their own way on the land.

Long before amphibians, when life originated on this planet, the first and simplest cells had to make more of themselves from the dead, raw materials around them. They produced the energy they needed from chemical reactions – reactions that would have taken place without them but which they could harness for their own use, most likely by linking them to the transport of electrically-charged particles across their own boundaries with the outside world. The supply of this chemical energy was limited, however, and the evolution of more complex and versatile forms of life could only come after one of those early cells had discovered an important trick – how to get its energy from the sun.

The heart of the green machine

In photosynthesis, light energy is converted into a chemical form to be used by a living cell for any one of its vital activities. Today we can see that the trick of photosynthesis has come a long way from its early beginnings, but we can still glimpse the essential features behind the elaborations that have come from three thousand million years or so of evolutionary change.

To store light energy, a living cell must use it to drive a photochemical reaction. The earliest chemical event in photosynthesis is in fact a photoelectrochemical reaction of the green pigment, chlorophyll. The chlorophyll molecule, after excitation with a quantum of light, returns to its original state by way of a chemical change in which it loses an electron to a neighbouring molecule of a closely related substance called phaeophytin. The phaeophytin hands the electron on to a quinone molecule, and a series of other molecules in turn

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play 'pass the electron', a game in which energy is transformed again – this time from electrical potential to stored chemical energy in the making of a molecule known as ATP. This last step uses precisely the same mechanism our own cells use when we get energy from our food – in both cases the energy-rich molecule ATP is built up from its components because electrically charged particles are moved from outside to inside, across the membrane boundary of the cell.

In 1985 three German biochemists reported on a structure obtained from X-ray diffraction studies of crystals of a pure substance obtained from the photosynthetic membranes of purple bacteria. The structure showed chlorophyll, phaeophytin and quinone molecules held in a very precise symmetrical arrangement by two proteins, each of which must cross the photosynthetic membrane five times. From this structure, depicted in figure 1, we can see for the first time the molecular machine that carries out the crucial, early events of photosynthesis. This work earned Michel, Deisenhofer and Huber the 1988 Nobel Prize for Chemistry. One of the many delights of this work is that it shows that movement of electrical charge really does take place from one side of the boundary membrane to the other, a conjecture of the British scientist Peter Mitchell that formed part of his Nobel-prizewinning theory of how cells transform energy.

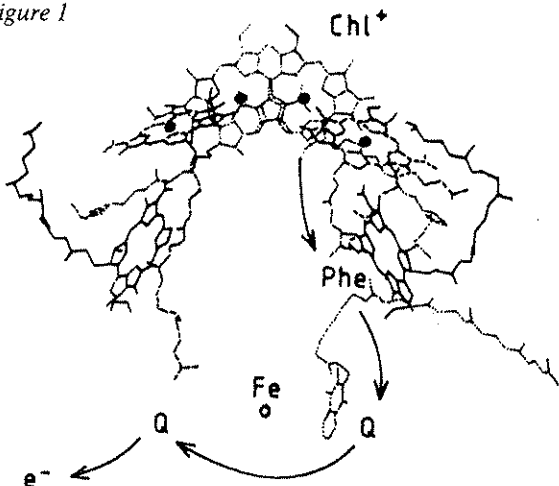
Harvesting the Sun

The beautiful structure of the reaction centre of photosynthesis is not, however, the whole story. Before the reaction centre chlorophyll molecule can be excited, the light energy must be harvested by a network of other chlorophyll molecules, and the energy must pass among them before it can reach the reaction centre. The light-harvesting chlorophylls must be held by protein so that they fit into the reaction centre in some way. Such chlorophylls act rather like a funnel for light energy with the reaction centre sitting under the hole at the bottom, as shown in figure 2.

The operation of the photosynthetic apparatus depends critically on a balance between its different components. The antenna of chlorophylls and proteins that harvest light must be large and operate at high efficiency when sunlight is in short supply – but a sudden, bright shaft of light will then dangerously overload the energy converting reaction centre unit if there is no way of matching the supply of light energy with the demand that the reaction centre can sustain. The rapid fluctuations in supply of light energy are easily seen in woodland on a sunny day, a snapshot of which is shown in figure 3. Plants, algae and photosynthetic bacteria clearly need to be adaptable to changes in intensity of the light they use.

When light is in short supply, photosynthesis must operate at maximum efficiency. This means that the wasted light energy must be reduced to a minimum. In order to do this, the light energy needs to be distributed between the energy-converting reaction centres in direct proportion to how fast each centre can work. This depends on

Figure 1



The structure of the main components of the photosynthetic reaction centre of the purple bacterium *Rhodospseudomonas viridis*, as first indicated by the work of Deisenhofer and co-workers. There are four chlorophyll molecules (Chl) (the dot at the centre of each chlorophyll ring is an atom of magnesium), two phaeophytin molecules (Phe), two quinone molecules (Q) (one not shown), and an iron atom (Fe). The first and fundamental step of energy conversion in photosynthesis is the movement of an electron (e^-) from one of the chlorophylls to the adjacent phaeophytin. This takes several millionths of a millionth of a second. The electron then moves first to one (in just less than a millionth of a second) then the other (in just less than a thousandth of a second) of the quinone molecules. For the hard-line reductionist the rest – life, history, Beethoven quartets, the UFC – is then just a question of more time and a little genetics. The structure shows that the electron must move across the boundary of the cell, from outside to inside, just as Peter Mitchell's theory predicted.

